

Rauhut-Currier Reaction of Nitroalkenes with Vinyl Sulfones

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Supporting Information

ABSTRACT: The Rauhut-Currier reaction between nitroalkenes and vinyl sulfones has been successfully carried out for the first time. The reaction proceeds in the presence of an imidazole/LiCl catalyst system in dioxane to provide the adducts possessing a synthetically useful 1-bis-sulfonyl-3-nitroalkene moiety in good to excellent yield. A one-pot transformation of the products to β -sulfonyl- α , β -unsaturated ketoximes has also been demonstrated.

he Morita-Baylis-Hillman (MBH) reaction has emerged as an efficient method for α -functionalization of activated alkenes with various electrophiles such as aldehydes, imines, etc. in the presence of an amine or phosphine catalyst.1 However, dimerization of the substrate in the absence of an electrophile was reported by Rauhut and Currier, prior to the MBH reaction, but was overshadowed by the enormous scope and applicability of the latter. Nevertheless, the Rauhut-Currier (RC) dimerizations of activated alkenes such as acrylonitrile, 3,4 vinyl ketones, 5,6 acrylates, 4,6 and vinyl sulfone and vinyl sulfonate⁶ have been reported. The scope of the reaction was further expanded by cross-coupling of different activated alkenes⁷ and by developing intramolecular and asymmetric versions.8 Applications of the RC reaction as the key step in natural product synthesis have also been investigated.

In recent years, nitroalkenes have been employed as substrates in the MBH¹⁰ and the RC¹¹⁻¹⁴ reactions.¹⁵ The RC cross-coupling of nitroalkenes with azodicarboxylates, 11 methyl vinyl ketone, 12,13 and acrylate 12,14 afforded synthetically useful multifunctional adducts. 15 The RC dimerization of nitroalkenes and nitrodienes proceeded in low to moderate yield in the presence of imidazole and hydroquinone. 13 However, the RC cross-coupling of nitroalkenes with vinyl sulfones remained unreported hitherto.

From another perspective, organosulfones exhibit a variety of biological properties, for instance, the antiprostate cancer drug Bicalutamide and the antileprosy drug Dapsone contain a sulfonyl group. Although similar biological and synthetic applications of sulfones have been extensively investigated, 16 the reactivity of vinyl bis-sulfones, in particular, as Michael acceptors has not received sufficient attention.¹⁷ Herein, we report a high-yielding RC coupling of nitroalkenes and nitrodienes with vinyl bis-sulfones in the presence of an imidazole/LiCl catalyst system.

Initially, we screened the classical MBH catalyst DABCO for the model reaction of nitroalkene 1a with sulfone 2a, which led to no conversion even after 48 h (Table 1, entry 1). While DMAP in CH₃CN gave the product 3a in 20% yield (entry 2)

Table 1. Optimization of the Reaction Conditions^a

				$_{/}$ NO $_{2}$
> NO	PhO ₂ S SO ₂ Ph	Catalyst, additive solvent	Λι ,	_SO₂Ph
Ar NO ₂ +	111020 002111	solvent		30 ₂ F11
1a	ll 2a	$Ar = 4-MeOC_6H_4$		SO₂Ph

entry	catalyst (equiv)	additive (equiv)	solvent	time (h)	% yield ^b
1	DABCO (1)		THF	48	с
2	DMAP (1)		CH ₃ CN	24	20^d
3	imidazole (1)		THF	16	44 ^d
4	imidazole (1)		ether	15	33^d
5	imidazole (1)		acetone	15	35^d
6	imidazole (1)		EtOAc	14	58
7	imidazole (1)		dioxane	14	50
8	imidazole (1)	LiCl (1)	THF	12	60
9	imidazole (1)	LiCl (1)	EtOAc	13	63
10	imidazole (1)	LiCl (1)	dioxane	12	88
11	imidazole (1)	LiCl (0.5)	dioxane	12	81
12	imidazole (0.5)	LiCl (1)	dioxane	24	83
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^a1a and 2a were used in 2:1 ratio. ^bIsolated yield after purification by silica gel chromatography. ^cNo reaction. ^dSubstantial quantity of 1a was recovered and 2a polymerized.

after 24 h, imidazole-catalyzed reactions in different solvents were much superior, and the product 3a was isolated in much higher yield (33-58%) in shorter reaction time (14-24 h, entries 3-7). Finally, we explored the possibility of using LiCl as a salting out agent and a weak Lewis acid. 12 This improved the yields only marginally (60-63%) in solvents such as THF and EtOAc (12-13 h, entries 8 and 9). However, complete conversion in 12 h with remarkable increase in the yield (88%) was observed when dioxane was used as the solvent (entry 10). Imidazole and LiCl in stoichiometric amounts were necessary to obtain the best yield in the shortest possible time (entry 10). While lowering the quantity of LiCl led to a drop in the yield,

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though marginal (81%, entry 11), a decrease in reaction rate with marginal decrease in the yield was observed upon lowering the quantity of imidazole (24 h, 83%, entry 12). It appears that the solvation of LiCl is minimized as the dielectric constant of the solvent decreases (ε : THF 7.52, EtOAc 6.08, dioxane 2.22) resulting in substantial improvement in the yield (entries 8–10).

The above optimized conditions were employed to investigate the scope of the reaction by screening different nitroalkenes 1 (Table 2). Thus nitroalkenes 1a-e with

Table 2. Scope of the Reaction^a

entry	R	3	time (h)	% yield ^b
1	4-OMeC ₆ H ₄	3a	12	88
2	$3,4-(OMe)_2C_6H_3$	3b	13	85
3	benzo $[d][1,3]$ dioxole	3c	15	82
4	$4-NMe_2C_6H_4$	3d	12	72
5	$4-MeC_6H_4$	3e	14	78
6	C_6H_5	3f	14	68
7	3-BrC ₆ H ₄	3g	18	60
8	2-ClC ₆ H ₄	3h	20	57
9	$4-FC_6H_4$	3i	12	72
10	$4-NO_2C_6H_4$	3j	18	61
11	2-furyl	3k	13	80
12	2-thienyl	31	13	86
13	$C_6H_5CH=CH$	3m	15	75
14	$2-NO_2C_6H_4CH=CH$	3n	17	83
15	cyclohexyl	30	15	61
16	$CH_3(CH_2)_3$	3p	15	62

^aCarried out at 0.5 mmol scale of sulfone 2a with 2 equiv of nitroalkene 1 and 1 equiv each of imidazole and LiCl. ^bIsolated yield after purification by silica gel chromatography.

electron-donating substituents on the aromatic ring provided the RC adducts $3\mathbf{a} - \mathbf{e}$ in high yield (72–88%, entries 1–5). The yields were lower (60–72%) with nitrostyrene $1\mathbf{f}$ and nitroalkenes $1\mathbf{g} - \mathbf{j}$ possessing weakly and strongly deactivating substituents on the aromatic ring (entries 6–10). The RC adducts $3\mathbf{k} - \mathbf{l}$ were isolated in excellent yield (80–86%) when heteroaromatic nitroalkenes $1\mathbf{k} - \mathbf{l}$ were reacted with sulfone $2\mathbf{a}$ (entries 11 and 12). While the yields of RC adducts $3\mathbf{m} - \mathbf{n}$ were high (75–83%) with nitrodienes $1\mathbf{m} - \mathbf{n}$ (entries 13 and 14), aliphatic nitroalkenes $1\mathbf{o} - \mathbf{p}$ afforded the products $3\mathbf{o} - \mathbf{p}$ in lower yield (61–62%, entries 15 and 16).

Under the above conditions, the reaction of a representative nitroalkene 1a with vinyl sulfone 2b afforded the desired product 4 in only low yield (25%, Scheme 1).

The structure and stereochemistry of RC adducts 3 were established by extensive NMR analysis (Figure 1). Appearance of the 1 H β to the nitro group in the RC adducts 3a–l and 4 in the range of 7.90–8.30 confirmed that the geometry of the double bond is *E*. The *E*,*E* geometry in the RC adducts 3m–n

Scheme 1

Ar NO₂ + SO₂Ph
$$\frac{\text{Imidazole, LiCl}}{\text{dioxane, RT}}$$
 Ar NO₂ $\frac{\text{NO}_2}{\text{SO}_2\text{Ph}}$ Ar = 4-MeOC₆H₄

H2: δ 7.19 (dd, J = 15.1, 11.7 Hz) H4: δ 7.13 (d, J = 7.0 Hz)

Figure 1. Structure and stereochemical assignment of 3 by NMR.

was evident from the ¹H-¹H NOESY spectrum of 3n. First of all, the three olefinic protons in 3n were assigned by ¹H-¹H COSY experiment. While H2 appeared as a dd at δ 7.19, H1 overlapped with aromatic protons in the range δ 7.67–7.75, and H3, the 1 H β to the nitro group, overlapped with aromatic protons in the range δ 7.82–7.89. Since H1 and H3 are very close in terms of chemical shifts and the corresponding signals overlap with the aromatic protons, their NOE cross peaks are not very diagnostic. However, the presence of strong NOE for H2 with H5 and H6 and a J value of 15.1 and 11.7 Hz for H2 supported the above assignment. The double bond geometry in RC adducts 3o-p was also determined by ¹H-¹H NOESY analysis. The absence of any NOE for H4 with H2 or H1 together with the presence of medium NOE for cyclohexyl protons with H2 and H1 confirmed the E geometry in 30. This assignment was unambiguously established by single crystal Xray analysis of a representative compound 3n (see the Experimental Section and the Supporting Information).

The proposed role of imidazole and LiCl in mediating the RC reaction of nitroalkenes 1 with vinyl sulfone 2a is outlined in Scheme 2. The initial conjugate addition of imidazole, a

Scheme 2. Proposed Mechanism for the Imidazole/LiCl Mediated Rauhut-Currier Reaction

nucleophilic Lewis base, is facilitated and the reversibility of the reaction is minimized by co-ordination of the nitro group with LiCl, a weak Lewis acid. This stabilized nitronate I then adds in a Michael fashion to vinyl sulfone 2a which is also activated by LiCl to generate intermediate II. The β -elimination of imidazole from II, triggered by the γ -hydrogen abstraction by the alkanesulfonyl group, completes the reaction to deliver the RC adduct 3.

Having developed an efficient method for the coupling of nitroalkenes 1 with vinyl sulfones 2 under the RC conditions, we sought to investigate the possible applications of the RC adducts 3. Thus, representative RC adducts 3e and 3i were subjected to metal/acid reduction under standard conditions (Scheme 3). Much to our amusement, the nitro group reduction stopped at the oxime stage, and elimination of one of the sulfonyl groups took place, all in one pot, to afford β -sulfonyl α,β -unsaturated oximes 5e and 5i in satisfactory yield

Scheme 3

(48–50%) as an inseparable mixture of syn/anti or anti/syn isomers in 86:14 to 89:11 ratio. The plausible mechanism for the transformation of the RC adduct 3 to oxime 5 involves formation of hydroxylamine III, its tautomerization to oxime IV, and elimination of benzenesulfinic acid from IV. The J values for the olefinic protons in anti and syn isomers were in the same range (15.5–15.9), as expected, indicating E geometry for the double bond in 5, which was further confirmed by NOE experiment. A moderate NOE between CH_2 and the proton β to the oxime group confirmed this assignment.

In conclusion, cross-coupling of nitroalkenes with vinyl sulfone has been carried out for the first time under the Rauhut–Currier conditions to afford the adducts in good to excellent yield in most cases. While vinyl bis-sulfones are good coupling partners, vinyl monosulfones were less efficient. A one-pot nitro group reduction—sulfone elimination under metal-acid conditions delivered synthetically useful β -sulfonyl α,β -unsaturated oximes in satisfactory yield and >85:15 anti/syn selectivity.

■ EXPERIMENTAL SECTION

General. The melting points recorded are uncorrected. NMR spectra (1 H, 1 H decoupled 13 C, 13 C-APT, 1 H $^{-1}$ H COSY, and 1 H $^{-1}$ H NOESY) were recorded with TMS as the internal standard. The coupling constants (J values) are given in hertz. High resolution mass spectra were recorded under ESI Q-TOF conditions. X-ray data were collected on a diffractometer equipped with graphite monochromated Mo K α radiation. The structure was solved by direct methods shelxs97 and refined by full-matrix least-squares against F^2 using shelxl97 software.

General Procedure for the Rauhut–Currier Reaction. To a stirred solution of nitroalkene 1 (1 mmol) and vinyl sulfone 2 (0.5 mmol) in dioxane (5 mL) were added imidazole (34 mg, 0.5 mmol) and LiCl (21 mg, 0.5 mmol), and the resulting mixture was stirred until the completion of the reaction (monitored by TLC). Then the reaction mixture was concentrated in vacuo, and the crude residue was directly subjected to silica gel column chromatography (60–120 mesh, *n*-hexane/ethyl acetate, 40:60) to afford pure products 3.

1-((*E*)-4-(4-Methoxyphenyl)-3-nitro-1-(phenylsulfonyl)but-3-enylsulfonyl)benzene (3a). Yellow solid; yield 88% (214 mg); mp 158–160 °C; IR (KBr, cm⁻¹) 3209 (w), 2923 (w), 1644 (m), 1603 (s), 1520 (s), 1448 (m), 1334 (s), 1309 (s), 1262 (s), 1177 (s), 1156 (s), 1079 (m), 736 (s); 1 H NMR (400 MHz, CDCl₃) δ 3.84 (d, J = 7.0 Hz, 2H), 3.89 (s, 3H), 5.18 (t, J = 7.0 Hz, 1H), 7.00 (d, J = 8.8 Hz, 2H), 7.49–7.53 (m, 4H), 7.56 (d, J = 8.8 Hz, 2H), 7.66–7.70 (m, 2H), 7.80–7.82 (m, 4H), 8.12 (s, 1H); 13 C NMR (100 MHz, CDCl₃) δ 24.46, 55.66, 79.10, 115.02, 123.33, 129.36, 129.41, 132.70, 134.95, 137.78, 138.02, 142.77, 162.02; MS (ES+, Ar) m/z (rel intensity) 510 (MNa⁺, 31), 505 (29), 489 ([MH + 1]⁺, 7), 488 (MH⁺, 31), 444 (13), 443 (24), 442 (100); HRMS (ES+, Ar) calcd for C₂₃H₂₂NO₇S₂ (MH⁺) 488.0838, found 488.0827.

1-((E)-4-(3,4-Dimethoxyphenyl)-3-nitro-1-(phenylsulfonyl)-but-3-enylsulfonyl)benzene (3b). Yellow solid; yield 85% (220

mg); mp 176–178 °C; IR (KBr, cm⁻¹) 3225 (w), 2923 (w), 1645 (m), 1556 (w), 1520 (s), 1448 (m), 1331 (m), 1314 (m), 1268 (s), 1245 (m), 1148 (s), 1079 (w), 1021 (m), 737 (m);

¹H NMR (400 MHz, CDCl₃) δ 3.85 (d, J = 7.4 Hz, 2H), 3.94 (s, 3H), 3.95 (s, 3H), 5.25 (t, J = 7.4 Hz, 1H), 6.94 (d, J = 8.4 Hz, 1H), 7.16 (dd, J = 8.4, 1.2 Hz, 1H), 7.36 (d, J = 1.2 Hz, 1H), 7.51 (t, J = 7.6 Hz, 4H), 7.68 (t, J = 7.6 Hz, 2H), 7.81 (d, J = 7.6 Hz, 4H), 8.11 (s, 1H);

¹³C NMR (100 MHz, CDCl₃) δ 24.93, 56.19, 56.52, 78.95, 111.41, 112.41, 123.66, 125.69, 129.38 (×2), 134.97, 137.93, 138.47, 142.75, 149.65, 151.82; MS (ES +, Ar) m/z (rel intensity) 540 (MNa⁺, 100), 519 ([M + 2]⁺, 20), 518 (M⁺, 50); HRMS (ES+, Ar) calcd for $C_{24}H_{24}NO_8S_2$ (MH⁺) 518.0943, found 518.0942.

(*E*)-5-(2-Nitro-4,4-bis(phenylsulfonyl)but-1-enyl)benzo[*d*]-[1,3]dioxole (3c). Yellow solid; yield 82% (205 mg); mp 184–186 °C; IR (KBr, cm⁻¹) 3231 (w), 3066 (w), 2918 (w), 1643 (vs), 1504 (w), 1487 (w), 1449 (m), 1334 (s), 1311 (m), 1262 (m), 1157 (m), 1037 (m), 741 (m); ¹H NMR (400 MHz, CDCl₃) δ 3.81 (d, *J* = 7.3 Hz, 2H), 5.17 (t, *J* = 7.3 Hz, 1H), 6.08 (s, 2H), 6.90 (d, *J* = 8.0 Hz, 1H), 7.08–7.11 (m, 2H), 7.50–7.54 (m, 4H), 7.68 (tt, *J* = 6.4, 1.1 Hz, 2H), 7.82 (dd, *J* = 8.4, 1.1 Hz, 4H), 8.14 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 24.49, 79.20, 102.18, 109.33, 109.82, 124.96, 126.72, 129.41, 129.51, 135.02, 137.81, 138.08, 143.42, 148.80, 150.35; MS (ES+, Ar) m/z (rel intensity) 524 (MNa⁺, 40), 503 ([M + 2]⁺, 9), 502 (MH⁺, 35), 496 (56), 480 (38), 458 (93), 456 (100); HRMS (ES+, Ar) calcd for $C_{23}H_{20}NO_8S_2$ (MH⁺) 502.0630, found 502.0639.

(*E*)-*N*,*N*-Dimethyl-4-(2-nitro-4,4-bis(phenylsulfonyl)but-1-enyl)aniline (3d). Yellow solid; yield 72% (180 mg); mp 190–192 °C; IR (KBr, cm⁻¹) 3065 (w), 2925 (w), 1668 (w), 1598 (s), 1531 (w), 1448 (w), 1332 (s), 1311 (s), 1292 (m), 1159 (vs), 1078 (m), 749 (m), 732 (m); 1 H NMR (400 MHz, CDCl₃) δ 3.10 (s, 6H), 3.91 (d, J = 7.2 Hz, 2H), 5.27 (t, J = 7.2 Hz, 1H), 6.73 (d, J = 8.9 Hz, 2H), 7.50 (t, J = 7.6 Hz, 4H), 7.56 (d, J = 8.9 Hz, 2H), 7.66 (t, J = 7.6 Hz, 2H), 7.82–7.84 (m, 4H), 8.04 (s, 1H); 13 C NMR (100 MHz, CDCl₃) δ 24.98, 40.21, 79.11, 112.20, 117.88, 129.34, 129.41, 133.83, 134.83, 138.37, 138.80, 139.62, 152.51; MS (ES+, Ar) m/z (rel intensity) 523 (MNa⁺, 15), 502 ([MH + 1]⁺, 24), 501 (MH⁺, 80), 444 (22), 379 (33), 363 (100), 341 (52); HRMS (ES+, Ar) calcd for $C_{24}H_{25}N_2O_6S_2$ (MH⁺) 501.1154, found 501.1169.

1-((*E*)-3-Nitro-1-(phenylsulfonyl)-4-*p*-tolylbut-3-enylsulfonyl)benzene (3e). Yellow solid; yield 78% (184 mg); mp 156–158 °C; IR (KBr, cm⁻¹) 3225 (w), 3055 (w), 1650 (vs), 1604 (s), 1517 (w), 1505 (w), 1443 (w), 1328 (s), 1310 (vs), 1279 (m), 1156 (m), 1080 (w), 742 (m); ¹H NMR (400 MHz, CDCl₃) δ 2.44 (s, 3H), 3.81 (d, J = 7.2 Hz, 2H), 5.14 (t, J = 7.2 Hz, 1H), 7.29 (d, J = 8.0 Hz, 2H), 7.44 (d, J = 8.0 Hz, 2H), 7.47–7.51 (m, 4H), 7.65–7.69 (m, 2H), 7.78 (dd, J = 8.4, 1.1 Hz, 4H), 8.14 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 21.72, 24.26, 79.23, 128.31, 129.34, 129.47, 130.21, 130.34, 134.95, 137.65, 138.09, 141.69, 144.33; MS (ES+, Ar) m/z (rel intensity) 494 (MNa⁺, 10), 489 (16), 474 ([MH + 2]⁺, 13), 473 ([MH + 1]⁺, 29), 472 (MH⁺, 100); HRMS (ES+, Ar) calcd for C₂₃H₂₂NO₆S₂ (MH⁺) 472.0889, found 472.0887.

1-((*E*)-3-Nitro-4-phenyl-1-(phenylsulfonyl)but-3-enylsulfonyl)benzene (3f). Yellow solid; yield 68% (156 mg); mp 174–176 °C; IR (KBr, cm⁻¹) 3065 (w), 2930 (w), 1648 (m), 1558 (m), 1525 (s), 1448 (m), 1333 (vs), 1264 (m), 1157 (vs), 1079 (m), 737 (s); ¹H NMR (400 MHz, CDCl₃) δ 3.78 (d, J = 7.2 Hz, 2H), 5.13 (t, J = 7.2 Hz, 1H), 7.47–7.54 (m, 9H), 7.67 (t, J = 7.5 Hz, 2H), 7.77 (dd, J = 8.4, 1.1 Hz, 4H), 8.19 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 24.21, 79.32, 129.38, 129.49, 129.53, 130.08, 130.84, 131.32, 135.00, 137.55, 137.86, 145.32; MS (ES+, Ar) m/z (rel intensity) 480 (MNa⁺, 7), 475 (12), 460 ([MH + 2]⁺, 14), 459 ([MH + 1]⁺, 29), 458 (MH⁺, 100); HRMS (ES+, Ar) calcd for $C_{22}H_{20}NO_6S_2$ (MH⁺) 458.0732, found 458.0746.

1-(*E*)-4-(3-Bromophenyl)-3-nitro-1-(phenylsulfonyl)but-3-enylsulfonyl)benzene (3g). Yellow solid; yield 60% (161 mg); mp 122–124 °C; IR (KBr, cm⁻¹) 3065 (w), 2927 (w), 1653 (m), 1525 (s), 1448 (m), 1334 (vs), 1287 (m), 1156 (s), 1078 (s), 737 (s), 685 (s); ¹H NMR (400 MHz, CDCl₃) δ 3.72 (d, J = 7.2 Hz, 2H), 5.12 (t, J = 7.2 Hz, 1H), 7.34 (t, J = 7.8 Hz, 1H), 7.48–7.52 (m, 5H), 7.60 (d, J = 8.2 Hz, 1H), 7.63 (d, J = 0.8 Hz, 1H), 7.67 (td, J = 8.3, 0.8, 2H),

7.77 (dd, J = 8.3, 0.8 Hz, 4H), 8.08 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 24.22, 79.22, 123.33, 128.14, 129.41, 129.47, 130.94, 132.66, 133.33, 133.57, 135.08, 136.10, 137.39, 146.36; MS (ES+, Ar) m/z (rel intensity) 538 ([MH + 2]+, 23), 537 ([MH + 1]+, 100), 536 (MH+, 65); HRMS (ES+, Ar) calcd for $C_{22}H_{19}NO_6S_2Br$ (MH+) 535.9837, found 535.9855.

1-((*E*)-4-(2-Chlorophenyl)-3-nitro-1-(phenylsulfonyl)but-3-enylsulfonyl)benzene (3h). Yellow liquid; yield 57% (140 mg); IR (neat, cm⁻¹) 3063 (w), 2929 (w), 1657 (m), 1559 (w), 1528 (s), 1473 (w), 1447 (w), 1340 (vs), 1266 (w), 1157 (s), 1078 (m), 753 (s); 1 H NMR (400 MHz, CDCl₃) δ 3.68 (d, J = 7.3 Hz, 2H), 5.14 (t, J = 7.3 Hz, 1H), 7.49–7.55 (m, 4H), 7.66–7.75 (m, 4H), 7.75–7.80 (m, 4H), 8.30 (s, 1H), 8.31 (d, J = 8.4 Hz, 2H); 13 C NMR (100 MHz, CDCl₃) δ 24.13, 79.30, 127.59, 129.36, 129.42, 129.51, 129.71, 130.19, 130.25, 131.64, 134.85, 134.99, 137.43, 146.83; MS (ES+, Ar) m/z (rel intensity) 494 ([MH + 2]+, 40), 493 ([MH + 1]+, 25), 492 (MH+, 100), 297 (17); HRMS (ES+, Ar) calcd for $C_{22}H_{19}NO_6S_2Cl$ 492.0342, found 492.0337.

1-((*E*)-4-(4-Fluorophenyl)-3-nitro-1-(phenylsulfonyl)but-3-enylsulfonyl)benzene (3i). Yellow solid; yield 72% (171 mg); mp 116–118 °C; IR (KBr, cm⁻¹) 3056 (m), 2987 (w), 1655 (w), 1603 (w), 1564 (w), 1527 (m), 1510 (m), 1422 (w), 1336 (m), 1266 (s), 1161 (m), 745 (vs); ¹H NMR (400 MHz, CDCl₃) δ 3.74 (d, *J* = 7.3 Hz, 2H), 5.16 (t, *J* = 7.3 Hz, 1H), 7.16 (t, *J* = 8.6 Hz, 2H), 7.50 (t, *J* = 7.7 Hz, 4H), 7.56 (dd, *J* = 8.6, 5.3 Hz, 2H), 7.67 (t, *J* = 7.7 Hz, 2H), 7.78 (dd, *J* = 7.7, 0.8 Hz, 4H), 8.13 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 24.27, 79.06, 116.73 (d, *J*_{C-F} = 22.0 Hz), 127.37 (d, *J*_{C-F} = 5.0 Hz), 129.41 (d, *J*_{C-F} = 2.0 Hz), 132.40 (d, *J*_{C-F} = 8.0 Hz), 135.04, 136.82, 137.54, 145.01, 162.72, 165.24; MS (ES+, Ar) m/z (rel intensity) 476 (MH⁺, 100); HRMS (ES+, Ar) calcd for C₂₂H₁₉NO₆S₂F 476.0638, found 476.0627.

1-((*E*)-3-Nitro-4-(4-nitrophenyl)-1-(phenylsulfonyl)but-3-enylsulfonyl)benzene (3j). Yellow solid; yield 61% (153 mg); mp 158–160 °C; IR (KBr, cm⁻¹) 3066 (w), 2926 (w), 1657 (w), 1525 (s), 1448 (w), 1345 (vs), 1264 (w), 1157 (m), 1079 (w), 749 (m), 737 (m); 1 H NMR (400 MHz, CDCl₃) δ 3.69 (d, J = 7.5 Hz, 2H), 5.14 (t, J = 7.5 Hz, 1H), 7.49–7.53 (m, 4H), 7.67–7.73 (m, 3H), 7.76–7.78 (m, 4H), 8.21 (s, 1H), 8.31 (d, J = 8.8 Hz, 2H); 13 C NMR (100 MHz, CDCl₃) δ 24.41, 78.96, 124.45, 129.46, 129.52, 130.63, 135.22, 135.24, 137.43, 137.84, 147.77, 148.62; MS (ES+, Ar) m/z (rel intensity) 526 (MNa⁺, 8), 525 (29), 520 (38), 505 ([MH + 1]⁺, 16), 504 (MH⁺, 32), 503 (M⁺, 100), 297 (17); HRMS (ES+, Ar) calcd for $C_{22}H_{19}N_2O_8S_2Na$ 526.0481, found 526.0459.

(E)-2-(2-Nitro-4,4-bis(phenylsulfonyl)but-1-enyl)furan (3k). Yellow solid; yield 80% (179 mg); mp 126–128 °C; IR (KBr, cm⁻¹) 3152 (w), 3064 (m), 2927 (w), 1648 (s), 1560 (m), 1513 (s), 1448 (w), 1314 (vs), 1158 (vs), 1080 (m), 1024 (m), 928 (m), 753 (s); ¹H NMR (400 MHz, CDCl₃) δ 4.05 (d, J = 7.3 Hz, 2H), 5.18 (t, J = 7.3 Hz, 1H), 6.58 (dd, J = 3.5, 1.8 Hz, 1H), 6.96 (d, J = 3.5 Hz, 1H), 7.50–7.55 (m, 5H), 7.67 (t, J = 7.5 Hz, 2H), 7.86–7.87 (m, 4H), 7.89 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 24.70, 80.17, 113.48, 122.64, 123.79, 129.31, 129.50, 134.87, 138.01, 141.01, 146.82, 147.73; MS (ES+, Ar) m/z (rel intensity) 449 ([MH + 1]⁺, 32), 448 (MH⁺, 100), 447 (12), 402 (44), 398 (16); HRMS (ES+, Ar) calcd for $C_{20}H_{18}NO_2S_2$ (MH⁺) 448.0525, found 448.0533.

(*E*)-2-(2-Nitro-4,4-bis(phenylsulfonyl)but-1-enyl)thiophene (3l). Yellow solid; yield 86% (200 mg); mp 134–136 °C; IR (KBr, cm⁻¹) 3225 (w), 3060 (w), 2944 (w), 1642 (s), 1583 (w), 1503 (m), 1495 (m), 1449 (w), 1402 (w), 1325 (s), 1303 (vs), 1154 (s), 1082 (m), 732 (s); ¹H NMR (400 MHz, CDCl₃) δ 3.89 (d, J = 7.3 Hz, 2H), 5.27 (t, J = 7.3 Hz, 1H), 7.19 (dd, J = 5.0, 3.7 Hz, 1H), 7.52 (td, J = 8.1, 1.0 Hz, 4H), 7.57 (d, J = 3.7 Hz, 1H), 7.66–7.70 (m, 3H), 7.86 (dd, J = 8.1, 1.0 Hz, 4H), 8.29 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 25.26, 79.10, 128.65, 129.20, 129.39, 131.26, 133.62, 133.69, 134.97, 136.62, 137.87, 141.12; MS (ES+, Ar) m/z (rel intensity) 466 ([MH + 2]+, 18), 465 ([MH + 1]+, 26), 464 (MH+, 100), 418 (22); HRMS (ES+, Ar) calcd for C₂₀H₁₈NO₆S₃ (MH+) 464.0296, found 464.0285.

1-((1*E*,3*E*)-4-Nitro-6,6-bis(phenylsulfonyl)hexa-1,3-dienyl)benzene (3m). Yellow solid; yield 75% (181 mg); mp 206–208 °C; IR (KBr, cm⁻¹) 3055 (w), 2923 (w), 1629 (vs), 1496 (m), 1445 (w),

1328 (s), 1306 (s), 1163 (s), 1148 (vs), 1079 (m), 754 (s), 733 (s); 1 H NMR (400 MHz, CDCl₃) δ 3.70 (d, J = 6.9 Hz, 2H), 5.12 (t, J = 6.9 Hz, 1H), 7.18 (ABq, J = 15.2 Hz, 2H), 7.40–7.45 (m, 3H), 7.52–7.56 (m, 4H), 7.57–7.60 (m, 2H), 7.69 (tt, J = 6.9, 1.1 Hz, 2H), 7.84 (dd, J = 8.5, 2.5 Hz, 1H), 7.87 (dd, J = 8.4, 1.1 Hz, 4H); 13 C NMR (100 MHz, CDCl₃) δ 24.17, 79.48, 121.21, 128.37, 129.25, 129.37, 129.47, 130.78, 135.02, 135.49, 138.13, 138.90, 142.59, 147.45; MS (ES+, Ar) m/z (rel intensity) 486 ([MH + 2]+, 13), 485 ([MH + 1]+, 27), 484 (MH+, 100), 296 (15); HRMS (ES+, Ar) calcd for $C_{24}H_{22}NO_6S_2$ (MH+) 484.0889, found 484.0874.

1-Nitro-2-((1E,3E)-4-nitro-6,6-bis(phenylsulfonyl)hexa-1,3dienyl)benzene (3n). Yellow solid; yield 83% (219 mg); mp 120-122 °C; IR (KBr, cm⁻¹) 3155 (m), 2988 (m), 1727 (w), 1618 (w), 1528 (s), 1446 (m), 1350 (s), 1244 (s), 1023 (vs), 982 (s), 785 (m), 737 (s); ¹H NMR (400 MHz, CDCl₃) δ 3.71 (d, J = 6.9 Hz, 2H), 5.10 (t, I = 6.9 Hz, 1H), 7.19 (dd, I = 15.1, 11.7 Hz, 1H), 7.52-7.58 (m, I)5H), 7.67-7.75 (m, 4H), 7.82-7.89 (m, 6H), 8.06 (dd, J = 8.2, 1.0 Hz, 1H); 13 C NMR (100 MHz, CDCl₃) δ 24.25, 79.23, 125.23, 125.99, 129.27, 129.57, 129.90, 130.53, 131.33, 134.04, 135.12, 137.72, 138.19, 141.25, 144.45, 148.17; MS (ES+, Ar) m/z (rel intensity) 531 $([MH + 2]^+, 15), 530 ([MH + 1]^+, 28), 529 (MH^+, 100); HRMS (ES$ +, Ar) calcd for $C_{24}H_{21}N_2O_8S_2$ (MH⁺) 529.0739, found 529.0765; $C_{52}H_{48}N_4O_{18}S_4$, M = 1145.18, Orthorhombic, space group *Pna21*, a = 13.2930 (3) Å, b = 11.5155 (3) Å, c = 34.5292 (9) Å, $\alpha = 90^{\circ}$, $\beta = 90^{\circ}$, $\gamma = 90^{\circ}$, V = 5285.6 (2) Å³, $D_c = 1.439$ Mg/m³, Z = 4, F(000) = 2384, $\lambda = 0.71073 \text{ Å}, \mu = 0.259 \text{ mm}^{-1}, \text{ total/unique reflections} = 46996/$ 9252 [R(int) = 0.0497], T = 150(2) K, θ range = 3.12–24.98°, final R $[I > 2\sigma(I)]$: R1= 0.0486, wR2 = 0.1185; R (all data): R1 = 0.0576, wR2 = 0.1270 (two molecules in the unit cell + EtOAc). Confirmed by ¹H-¹H COSY and NOESY experiments as well as single crystal X-ray analysis.

1-((*E*)-4-Cyclohexyl-3-nitro-1-(phenylsulfonyl)but-3-enylsulfonyl)benzene (3o). White solid; yield 61% (141 mg); mp 122–124 °C; IR (KBr, cm⁻¹) 2923 (m), 2846 (w), 1645 (vs), 1509 (m), 1446 (w), 1315 (vs), 1149 (s), 738 (m); ¹H NMR (400 MHz, CDCl₃) δ 1.15–1.28 (m, 3H), 1.32–1.42 (m, 2H), 1.70–1.78 (m, 5H), 2.57 (qt, J = 10.9, 3.8 Hz, 1H), 3.50 (d, J = 7.0 Hz, 2H), 5.04 (t, J = 7.0 Hz, 1H), 7.13 (d, J = 7.0 Hz, 1H), 7.52 (td, J = 7.9, 1.1 Hz, 4H), 7.67 (tt, J = 7.9, 1.1 Hz, 2H), 7.82 (dd, J = 7.9, 1.1 Hz, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 23.62, 25.07, 25.67, 31.56, 38.17, 79.51, 129.33, 129.44, 134.93, 138.10, 143.62, 146.87; MS (ES+, Ar) m/z (rel intensity) 466 ([MH + 2]+, 16), 465 ([MH + 1]+, 30), 464 (MH+, 100); HRMS (ES+, Ar) calcd for C₂₂H₂₆NO₆S₂ (MH+) 464.1202, found 464.1182. Confirmed by ¹H–¹H COSY and NOESY experiments.

1-((Z)-3-Nitro-1-(phenylsulfonyl)oct-3-enylsulfonyl)benzene (**3p).** White solid; yield 62% (135 mg); mp 102–104 °C; IR (KBr, cm⁻¹) 3066 (w), 2959 (m), 2934 (m), 1657 (m), 1657 (s), 1646 (s), 1520 (s), 1448 (m), 1334 (vs), 1311 (s), 1157 (vs), 1079 (m), 737 (s); ¹H NMR (400 MHz, CDCl₃) 0.91 (t, J = 7.4 Hz, 3H), 1.39 (sextet, J = 7.4 Hz, 2H), 1.51 (q, J = 7.4 Hz, 2H), 2.43 (q, J = 7.4 Hz, 2H), 3.49 (d, J = 7.0 Hz, 2H), 5.03 (t, J = 7.0 Hz, 1H), 7.31 (t, J = 7.4 Hz, 1H), 7.53 (t, J = 7.6 Hz, 4H), 7.68 (t, J = 7.6 Hz, 2H), 7.81–7.83 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 13.94, 22.62, 23.62, 28.59, 30.31, 79.26, 129.32, 129.44, 134.96, 138.06, 143.16, 144.87; MS (ES+, Ar) m/z (rel intensity) 440 ([MH + 2]+, 15), 439 ([MH + 1]+, 27), 438 (MH+, 100), 420 (15), 278 (54); HRMS (ES+, Ar) calcd for $C_{20}H_{24}NO_6S_2$ (MH+) 438.1045, found 438.1049.

(*E*)-1-Methoxy-4-(2-nitro-4-(phenylsulfonyl)but-1-enyl)benzene (4a). Yellow solid; yield 25% (44 mg); mp 116–118 °C; IR (KBr, cm⁻¹) 3060 (w), 2934 (w), 1645 (w), 1603 (s), 1519 (s), 1506 (s), 1446 (m), 1317 (m), 1262 (m), 1179 (s), 1143 (m); ¹H NMR (400 MHz, CDCl₃) δ 3.28–3.31 (m, 2H), 3.43–3.47 (m, 2H), 3.87 (s, 3H), 6.95 (d, J = 8.8 Hz, 2H), 7.54 (d, J = 8.8 Hz, 2H), 7.61 (td, J = 7.6, 1.4 Hz, 2H), 7.71 (tt, J = 7.6, 1.4 Hz, 1H), 7.94–7.96 (m, 2H), 8.11 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 22.00, 53.30, 55.71, 115.17, 123.40, 128.35, 129.73, 132.57, 134.34, 136.63, 138.55, 144.65, 162.15; MS (ES+, Ar) m/z (rel intensity) 370 (MNa⁺, 100); HRMS (ES+, Ar) calcd for C₁₇H₁₈NO₅SNa 371.0803, found 371.0800.

One-Pot Metal/Acid Reduction—Sulfone Elimination of RC Adducts 3. To a stirred solution of RC adduct 3 (0.2 mmol) in a mixture of MeOH (1.5 mL), water (0.5 mL), and conc HCl (0.2 mL) was added iron dust (35 mg, 0.6 mmol). The resulting reaction mixture was heated to reflux on a water bath. After completion of reaction (monitored by TLC, ~30 min), the mixture was cooled to ambient temperature and filtered through a Celite pad. The filtrate was concentrated *in vacuo*, and the residue was diluted with water (5 mL), basified with 30% NaOH (10 mL), and extracted with ethyl acetate (3 × 10 mL). The combined organic layers were washed with brine (5 mL) and concentrated *in vacuo*, and the residue was purified by silica gel column chromatography by eluting with 20% EtOAC-pet ether.

(2*E*,3*E*)-4-(Phenylsulfonyl)-1-*p*-tolylbut-3-en-2-one Oxime (5a). White solid; yield 48% (30 mg, ~86:14, inseparable); mp 126-128 °C; IR (KBr, cm⁻¹) 3391 (m), 3060 (w), 2924 (m), 2854 (w), 1608 (w), 1515 (w), 1448 (m), 1309 (vs), 1150 (vs), 1085 (s), 998 (m), 831 (m), 800 (m), 754 (m); ¹H NMR (CDCl₃, 400 MHz, major isomer) δ 2.28 (s, 3H), 3.83 (s, 2H), 6.65 (d, J = 15.6 Hz, 1H), 7.01 (ABq, J = 16.5 Hz, 4H), 7.45 (d, J = 15.6 Hz, 1H), 7.50–7.53 (m, 2H), 7.60–7.64 (m, 1H), 7.79–7.84 (m, 2H); ¹³C NMR (CDCl₃, 100 MHz, major isomer) δ 21.16, 30.33, 127.84, 128.33, 129.53, 129.64, 132.10, 132.37, 133.80, 136.57, 139.41, 140.06, 155.60; MS (ES+) m/z (rel intensity) 338 (MNa⁺, 27), 325 (100), 186 (21), 132 (10); HRMS (ES+) calcd for $C_{17}H_{17}NO_3SNa$ (MNa⁺) 338.0821, found 338.0821.

(2*E*,3*E*)-1-(4-Fluorophenyl)-4-(phenylsulfonyl)but-3-en-2-one Oxime (5b). White solid; yield 50% (32 mg, ~89:11, inseparable); mp 156–158 °C; IR (KBr, cm⁻¹) 3509 (s), 3064 (w), 2924 (w), 1604 (m), 1583 (m), 1510 (s), 1324 (s), 1226 (m), 1151 (vs), 1085 (m), 826 (m), 754 (m); ¹H NMR (CDCl₃, 400 MHz, major isomer) δ 3.83 (s, 2H), 6.64 (d, J = 15.6 Hz, 1H), 6.88–6.94 (m, 2H), 7.05–7.09 (m, 2H), 7.50–7.55 (m, 3H), 7.62–7.66 (m, 1H), 7.82–7.84 (m, 2H); ¹³C NMR (CDCl₃, 100 MHz, major isomer) δ 29.94, 115.83 (d, $J_{C-F} = 22.0$ Hz), 127.83, 129.62, 130.02 (d, $J_{C-F} = 8.0$ Hz), 130.89 (d, $J_{C-F} = 4.0$ Hz), 132.38, 133.98, 139.26, 139.91, 155.24, 161.86 (d, $J_{C-F} = 244.0$ Hz); MS (ES+) m/z (rel intensity) 320 (MH⁺, 100), 295 (24), 245 (31), 231 (33); HRMS (ES+) calcd for $C_{16}H_{15}FNO_3S$ (MH⁺) 320.0751, found 320.0739.

■ ASSOCIATED CONTENT

S Supporting Information

CIF file for **3n** and copies of NMR spectra for all the new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) (a) Morita, K.; Suzuki, Z.; Hirose, H. Bull. Chem. Soc. Jpn. 1968, 41, 2815. (b) Baylis, A. B.; Hillman, M. E. D. German Patent DE 445 2155113, 1972; Chem. Abstr. 1963, 58, 11224a. For a recent review: (c) Basavaiah, D.; Veeraraghavaiah, G. Chem. Soc. Rev. 2012, 41, 68. For a book: (d) The Chemistry of the Morita-Baylis-Hillman Reaction; Shi, M., Wang, F.-J., Zhao, M. X., Wei, Y., Eds.; The Royal Society of Chemistry: Cambridge, U.K., 2011; pp 561.
- (2) (a) Rauhut, M. M.; Currier, H. U.S. Patent 3074999, 1963; Chem. Abstr. 1963, 58, 11224a. For a comprehensive review: (b) Aroyan, C. E.; Dermenci, A.; Miller, S. J. Tetrahedron 2009, 65, 4069 and references therein. See also: (c) Methot, J. L.; Roush, W. R. Adv. Synth. Catal. 2004, 346, 1035.

- (3) (a) Mc Clure, J. D. U.S. Patent 3225083, 1965. (b) Baizer, M. M.; Anderson, J. D. *J. Org. Chem.* 1965, 30, 1357. (c) Basavaiah, D.; Gowriswari, V. V. L.; Dharma Rao, P.; Bharathi, T. K. *J. Chem. Res.* (S) 1995, 267
- (4) (a) Amri, H.; Rambaud, M.; Villieras, J. Tetrahedron Lett. 1989, 30, 7381. (b) Drewes, S. E.; Emslie, N. D.; Karodia, N. Synth. Commun. 1990, 20, 1915. (c) Jenner, G. Tetrahedron Lett. 2000, 41, 3091.
- (5) (a) Amri, H.; Villieras, J. Tetrahedron Lett. 1986, 27, 4307.
 (b) Basavaiah, D.; Gowriswari, V. V. L.; Bharathi, T. K. Tetrahedron Lett. 1987, 28, 4591. (c) Mc Dougal, S. E.; Schaus, S. E. Angew. Chem., Int. Ed. 2006, 45, 3117.
- (6) Kaye, P. T.; Nocanda, X. W. J. Chem. Soc., Perkin Trans. 1 2002, 1318.
- (7) Acrylate and acrylonitrile with fumaric/maleic esters: (a) Morita, K.; Kobayashi, T. Bull. Chem. Jpn. 1969, 42, 2732. Acrylate with acrylonitrile: (b) Mc Clure, J. D. J. Org. Chem. 1970, 35, 3045. Enones with acrylate, acrylonitrile, and vinyl sulfone: (c) Hwu, J. R.; Hakimelahi, G. H.; Chou, C.-T. Tetrahedron Lett. 1992, 33, 6469. The same authors proposed Michael addition of the dienoate generated from the β -alkyl enone by the base to the desired activated alkene as the pathway, rather than the RC pathway. (d) Enals with vinyl sulfone: Atienza, R. L.; Scheidt, K. A. Aust. J. Chem. 2011, 64, 1158. Acrylonitrile with α -halomethyl acrylate and vinyl ketone: (e) Basavaiah, D.; Kumaragurubaran, N.; Sharada, D. S. Tetrahedron Lett. 2001, 42, 85. Acrylate, acrylonitrile, and vinyl ketone with α bromomethyl acrylate: (f) Basavaiah, D.; Sharada, D. S.; Kumaragurubaran, N.; Reddy, R. M. J. Org. Chem. 2002, 67, 7135. Allenoate with enone: (g) Evans, C. A.; Miller, S. J. J. Am. Chem. Soc. 2003, 125, 12394. MVK and acrylate with dihalonaphthoquinones: (h) Lee, C. H.; Lee, K.-J. Synthesis 2004, 12, 1941.
- (8) Intramolecular, selected articles: (a) Brown, P. M.; Kappel, N.; Murphy, P. J.; Koles, S. J.; Hursthouse, M. B. *Tetrahedron* **2007**, 63, 1100. (b) Luis, A. L.; Krische, M. J. *Synthesis* **2004**, 15, 2579. (c) Thalji, R. K.; Roush, W. R. *J. Am. Chem. Soc.* **2005**, 127, 16778. (d) Siedel, F. O.; Gladysz, J. A. *Adv. Synth. Catal.* **2008**, 350, 2443. Intramolecular asymmetric: (e) Aroyan, C. E.; Miller, S. J. *J. Am. Chem. Soc.* **2007**, 129, 256.
- (9) For selected recent examples: (a) Webber, P.; Krische, M. J. J. Org. Chem. 2008, 73, 9379. (b) Winbush, S. M.; Mergott, D. J.; Roush, W. R. J. Org. Chem. 2008, 73, 1818. (c) Stark, L. M.; Pekari, K.; Sorensen, E. J. Proc. Natl. Acad. Sci. U.S.A. 2004, 101, 1064.
- (10) Formaldehyde: (a) Rastogi, N.; Namboothiri, I. N. N.; Cojocaru, M. Tetrahedron Lett. 2004, 45, 4745. (b) Mohan, R.; Rastogi, N.; Namboothiri, I. N. N.; Mobin, S. M.; Panda, D. Bioorg. Med. Chem. 2006, 14, 8073. Other carbonyl compounds: (c) Deb, I.; Dadwal, M.; Mobin, S. M.; Namboothiri, I. N. N. Org. Lett. 2006, 8, 1201. (d) Deb, I.; Shanbhag, P.; Mobin, S. M.; Namboothiri, I. N. N. Eur. J. Org. Chem. 2009, 4091. (e) Kuan, H.-H.; Reddy, R. J.; Chen, K. Tetrahedron 2010, 66, 9875. Imines: (f) Rastogi, N.; Mohan, R.; Panda, D.; Mobin, S. M.; Namboothiri, I. N. N. Org. Biomol. Chem. 2006, 4, 3211. (g) Rajesh, K.; Shanbhag, P.; Raghavendra, M.; Bhardwaj, P.; Namboothiri, I. N. N. Tetrahedron Lett. 2010, 51, 846.
- (11) Dadwal, M.; Mobin, S. M.; Namboothiri, I. N. N. Org. Biomol. Chem. 2006, 4, 2525.
- (12) Dadwal, M.; Mohan, R.; Panda, D.; Mobin, S. M.; Namboothiri, I. N. N. Chem. Commun. 2006, 338.
- (13) Shanbhag, P.; Nareddy, P. R.; Dadwal, M.; Mobin, S. M.; Namboothiri, I. N. N. Org. Biomol. Chem. 2010, 8, 4867.
- (14) Intramolecular: Wang, X.-F.; Peng, L.; An, J.; Li, C.; Yang, Q.-Q.; Lu, L.-Q.; Gu, F.-L.; Xiao, W.-J. Chem.—Eur. J. 2011, 17, 6484.
- (15) Review: Kaur, K.; Namboothiri, I. N. N. Chimia 2012, 66, 913. (16) Selected reviews/books: (a) Simpkins, N. S. Sulfones in Organic Synthesis; Pergamon Press: Oxford, 1993. (b) Back, T. G.; Clary, K. N.; Gao, D. Chem. Rev. 2010, 110, 4498.
- (17) (a) Jian, X.; Yun-Peng, L.; Yan-Ling, L.; Poh-Shen, W.; Teck-Peng, L. Org. Lett. 2011, 13, 876. (b) Natalia, B.; Nekane, R. A. A.; Guillem, V.; Xavier, C.; Albert, M.; Ramon, R. New J. Chem. 2010, 34, 1816. (c) Sarah, M.; Marju, M.; Kadri, K.; Tonis, K.; Alexandre, A. Org. Lett. 2006, 8, 2559.